

Heisenberg's Uncertainty Principle: It is not possible to measure simultaneously both the position and momentum of a microscopic particle with absolute accuracy.

Uncertainty relation:

Let's assume that mean square deviation is related to uncertainty.

Consider A is the Hermitian operator.

$$A = A^\dagger$$

Let

$$A = A - \langle A \rangle$$

$$\langle A \rangle = \langle A - \langle A \rangle \rangle$$

$$\langle A \rangle = \langle A \rangle - \langle A \rangle \quad (\because \langle \langle A \rangle \rangle = \langle A \rangle)$$

$$\langle A \rangle = 0$$

$$\langle A^2 \rangle = \langle (A - \langle A \rangle)^2 \rangle$$

$$\langle A^2 \rangle = \langle (A^2 + \langle A \rangle^2 - 2A\langle A \rangle) \rangle$$

$$\langle A^2 \rangle = \langle A^2 \rangle + \langle A \rangle^2 - 2\langle A \rangle \langle A \rangle$$

$$\langle A^2 \rangle = \langle A^2 \rangle + \langle A \rangle^2 - 2\langle A \rangle^2$$

$$\langle A^2 \rangle = \langle A^2 \rangle - \langle A \rangle^2 \quad \text{--- (1)}$$

$$\langle A^2 \rangle = (\Delta A)^2$$

uncertainty in A is $(\Delta A)^2 = \langle A^2 \rangle$ --- (A)

uncertainty in B is $(\Delta B)^2 = \langle B^2 \rangle$ --- (B)

$$A = A - \langle A \rangle$$

$$B = B - \langle B \rangle$$

A and B are the properties of uncertainty

let us assume another operator which is not Hermitian.

$$O = A + i\lambda B$$

$$O^\dagger = A - i\lambda B$$

$$(\because O \neq O^\dagger)$$

assume system is stated by ψ

$$O\psi = \phi \quad \text{--- (2)}$$

$$\langle A \rangle = (\phi | A | \phi) = \int \phi^* A \phi d\tau \geq 0$$

$$(O\psi, O\psi) = (\psi, O^\dagger O \psi)$$

$$(O\psi, O\psi) = (\psi, (A - i\lambda B)(A + i\lambda B)\psi) \quad \text{--- (3)}$$

$$\langle A \rangle = (\psi, A\psi)$$

$$\langle (A - i\lambda B)(A + i\lambda B) \rangle \geq 0$$

$$\langle A^2 + \lambda^2 B^2 + i\lambda (AB - BA) \rangle \geq 0$$

$$\langle A^2 + \lambda^2 B^2 + i\lambda [A, B] \rangle \geq 0$$

$$\underbrace{\langle A^2 \rangle + \lambda^2 \langle B^2 \rangle + i\lambda \langle [A, B] \rangle}_{\text{H.C.}} \geq 0 \quad \text{--- (4)}$$

$$f(\lambda) = \langle A^2 \rangle + \lambda^2 \langle B^2 \rangle + i\lambda \langle [A, B] \rangle$$

$f(\lambda)$ should be small as possible

$$\frac{df(\lambda)}{d\lambda} = 0$$

$$2\lambda \langle B^2 \rangle + i \langle [A, B] \rangle = 0$$

$$\lambda = -\frac{i \langle [A, B] \rangle}{2 \langle B^2 \rangle}$$

Put the value of λ in equation (1).

$$\langle A^2 \rangle - \frac{\langle [A, B] \rangle^2}{4 \langle B^2 \rangle} + \frac{\langle [A, B] \rangle^2}{2 \langle B^2 \rangle} \geq 0$$

$$\langle A^2 \rangle - \frac{\langle [A, B] \rangle^2}{4 \langle B^2 \rangle} + \frac{2 \langle [A, B] \rangle^2}{4 \langle B^2 \rangle} \geq 0$$

$$\langle A^2 \rangle + \frac{\langle [A, B] \rangle^2}{4 \langle B^2 \rangle} \geq 0$$

$$\langle A^2 \rangle \geq -\frac{\langle [A, B] \rangle^2}{4 \langle B^2 \rangle}$$

$$\langle A^2 \rangle \langle B^2 \rangle \geq -\frac{\langle [A, B] \rangle^2}{4}$$

$$\boxed{\langle \Delta A \rangle^2 \langle \Delta B \rangle^2 \geq -\frac{\langle [A, B] \rangle^2}{4}}$$

from equation
(A) and (B)

let $A = x$ position

$B = p_x$ momentum, $p_x = -i\hbar \frac{\partial}{\partial x}$

$$(\Delta x)^2 (\Delta p_x)^2 \geq - \frac{(\langle [x, p_x] \rangle)^2}{4}$$

$$[x, p_x] = -i\hbar$$

$$(\Delta x)^2 (\Delta p_x)^2 \geq - \frac{(\langle -i\hbar \rangle)^2}{4}$$

$$(\Delta x)^2 (\Delta p_x)^2 \geq - \frac{\hbar^2}{4} \quad (\epsilon^2 = -1)$$

$$(\Delta x)^2 (\Delta p_x)^2 \geq \frac{\hbar^2}{4}$$

$$\boxed{\Delta x \Delta p_x \geq \frac{\hbar}{2}} \quad \left(\hbar = \frac{h}{2\pi} \right)$$

$h \rightarrow$ Planck's constant

This is called uncertainty principle.

If Δx is small then Δp_x will be very large.

Uncertainty
in momentum

If Δp_x is small then Δx is very large
(uncertainty in position).

we cannot see a microscopic particle without disturbing it.

$$\Delta y \Delta p_y \geq \frac{\hbar}{2}, \quad \Delta z \Delta p_z \geq \frac{\hbar}{2}.$$

or

$$\Delta E \Delta t \geq \frac{\hbar}{2}$$